

C17	0.6731 (6)	0.5120 (5)	0.8071 (3)	0.0753 (18)
C18	0.5215 (5)	0.5211 (4)	0.8068 (3)	0.0561 (15)
C1	0.1058 (4)	0.3083 (3)	0.6678 (2)	0.0423 (11)
C6	0.2221 (6)	0.3510 (4)	0.6278 (3)	0.0561 (15)
C5	0.2054 (8)	0.3713 (4)	0.5530 (3)	0.0703 (18)
C4	0.0770 (9)	0.3493 (4)	0.5189 (3)	0.0847 (25)
C3	-0.0376 (8)	0.3065 (5)	0.5572 (3)	0.0803 (21)
C2	-0.0254 (6)	0.2864 (4)	0.6319 (3)	0.0589 (14)

Table 2. Selected geometric parameters (Å, °)

N1—C7	1.320 (5)	C12—C13	1.483 (5)
N1—C9	1.373 (5)	N3—O	1.371 (4)
C7—N2	1.371 (5)	C13—C14	1.389 (6)
C7—C1	1.470 (5)	C13—C18	1.388 (6)
N2—C8	1.394 (5)	C14—C15	1.397 (6)
N2—C12	1.433 (4)	C15—C16	1.352 (8)
C8—C9	1.369 (5)	C16—C17	1.391 (9)
C8—C11	1.487 (6)	C17—C18	1.393 (7)
C9—C10	1.491 (6)	C1—C6	1.392 (6)
C11—C19	1.519 (6)	C1—C2	1.392 (6)
C19—C20	1.387 (7)	C6—C5	1.389 (7)
C19—C24	1.372 (7)	C5—C4	1.35 (1)
C20—C21	1.375 (9)	C4—C3	1.37 (1)
C21—C22	1.37 (1)	C3—C2	1.383 (8)
C22—C23	1.37 (1)	C12—N3	1.261 (4)
C23—C24	1.38 (1)		
C7—N1—C9	107.6 (3)	C21—C22—C23	119.4 (7)
N1—C7—C1	123.8 (3)	C22—C23—C24	120.2 (6)
N1—C7—N2	109.8 (3)	C19—C24—C23	121.1 (5)
N2—C7—C1	126.3 (3)	N2—C12—C13	116.9 (3)
C7—N2—C12	127.5 (3)	N2—C12—N3	122.6 (3)
C7—N2—C8	107.6 (3)	N3—C12—C13	120.5 (3)
C8—N2—C12	124.6 (3)	C12—N3—O	113.1 (3)
N2—C8—C11	124.6 (3)	C12—C13—C18	120.8 (3)
N2—C8—C9	105.5 (3)	C12—C13—C14	120.0 (3)
C9—C8—C11	129.8 (4)	C14—C13—C18	119.2 (4)
N1—C9—C8	109.6 (3)	C13—C14—C15	120.8 (4)
C8—C9—C10	127.6 (4)	C14—C15—C16	119.8 (5)
N1—C9—C10	122.8 (4)	C15—C16—C17	120.3 (5)
C8—C11—C19	113.7 (4)	C16—C17—C18	120.6 (5)
C11—C19—C24	120.8 (4)	C13—C18—C17	119.4 (5)
C11—C19—C20	121.0 (4)	C7—C1—C2	117.9 (4)
C20—C19—C24	118.1 (5)	C7—C1—C6	122.9 (4)
C19—C20—C21	120.7 (5)	C6—C1—C2	119.2 (4)
C20—C21—C22	120.5 (7)	C1—C6—C5	119.6 (5)
C6—C5—C4	120.4 (5)	C4—C3—C2	120.4 (6)
C5—C4—C3	120.8 (5)	C1—C2—C3	119.7 (5)
N1—C7—C1—C2	32.9 (6)	C8—N2—C12—C13	76.2 (4)
N1—C7—N2—C8	-0.8 (4)	N2—C12—C13—C18	-159.0 (4)
C1—C7—N2—C12	5.7 (6)	N2—C12—N3—O	0.9 (5)
C7—N2—C12—N3	67.5 (5)	N3—C12—C13—C14	-156.6 (4)
C7—N2—C12—C13	-111.2 (4)	N3—C12—C13—C18	22.3 (6)
C7—N2—C8—C11	-174.3 (4)	C13—C12—N3—O	179.5 (3)
C8—N2—C12—N3	-105.1 (4)		

Intensity data were corrected for Lorentz and polarization effects. The structure was solved by direct methods with the *MULTAN80* system (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980); subsequent calculations were performed mainly using the *SHELX76* (Sheldrick, 1976) and *PARST* (Nardelli, 1983) systems of programs on the VAX3400 computer at the Centro Interdipartimentale di servizi per la Diffattometria a Raggi-X dell'Università di Messina. The structure was refined anisotropically by full-matrix least-squares methods.

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71464 (13 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: NA1035]

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## 2-Hydroxy-4,4-dimethyl-2-(2-methylphenyl)-morpholinium Bromide, C<sub>13</sub>H<sub>20</sub>NO<sub>2</sub><sup>+</sup>·Br<sup>-</sup>

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## Abstract

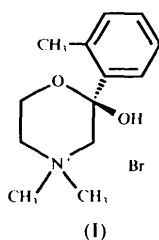
The morpholinium ring adopts a chair conformation with endocyclic torsion angles between 48.3 (4) and 60.7 (5)°. The hydroxyl group is in the axial position of the morpholinium ring, with a C—OH bond distance of 1.406 (5) Å. The hydroxy H atom points towards the Br ion; the interaction has an O⋯Br distance of 3.241 (3) Å, an H⋯Br distance of 2.35 (4) Å and an angle at H of 172 (4)°.

## Comment

As part of a larger study on structural changes in reactions involving tetrahedral intermediates as related to substituent effects in the structures of 2-substituted 2-hydroxy-4,4-dimethylmorpholinium salts (Garcia, 1986), the syntheses of some morpholinium bromides (Garcia-Guajardo, Fronczek &

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Gandour, 1986; Altbach, Fronczek, Gandour, Garcia, Lin & Watkins, 1988; Garcia, Fronczek & Gandour, 1992) were required. 2-Hydroxy-2-(2-methylphenyl)-4,4-dimethylmorpholinium bromide was prepared by condensing 2-methyl-2'-acetophenone with *N,N*-dimethylethanolamine following the procedure described by Garcia (1986). The colorless prisms had an m.p. of 435–437 K. Structural data for 2-(4-cyanophenyl)-2-hydroxy-4,4-dimethylmorpholinium bromide (Altbach *et al.*, 1988), 2-hydroxy-4,4-dimethyl-2-(4-tolyl)morpholinium bromide (Garcia *et al.*, 1992), 2-hydroxy-4,4,6-trimethyl-2-(4-nitrophenyl)morpholinium bromide (Garcia, Fronczek & Gandour, 1993) and 4,4-dimethyl-2-oxomorpholinium bromide (Garcia-Guajardo *et al.*, 1986) are in agreement with those of the title compound.



The morpholinium ring adopts a chair conformation with the torsion angles  $\omega_1$  (O1—C2—C1—N) =  $-50.3$  (4),  $\omega_2$  (C2—C1—N—C4) =  $48.3$  (4),  $\omega_3$  (C1—N—C4—C3) =  $-52.5$  (4),  $\omega_4$  (N—C4—C3—O1) =  $59.5$  (5),  $\omega_5$  (C4—C3—O1—C2) =  $-60.7$  (5) and  $\omega_6$  (C3—O1—C2—C1) =  $55.5$  (4)', which are distorted from the ideal chair conformation  $\bar{3}$  ( $D_{3d}$ )

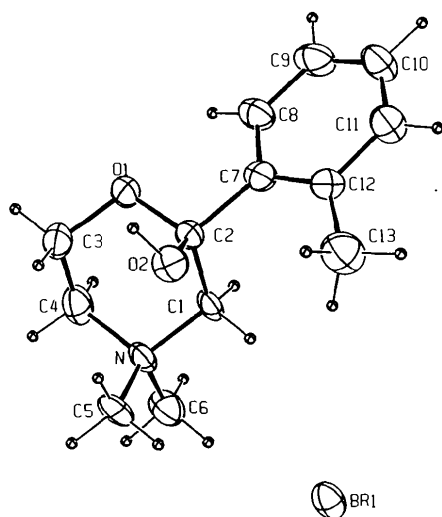


Fig. 1. ORTEP drawing (Johnson, 1965) of the molecule, representing heavy atoms as 40% probability ellipsoids and H atoms as circles of arbitrary size.

of cyclohexane (Hargittai & Hargittai, 1986), having torsion angles of alternating  $\pm 54.4^\circ$  (Hendrickson, 1967). The pharmacological activities of related compounds are reported by Anderson, Corey, Force, Jensen, Matz & Rivard (1966) and Lee, Stelly, Colucci, Garcia, Gandour & Quinn (1992).

## Experimental

### Crystal data

$C_{13}H_{20}NO_2 \cdot Br^-$

$M_r = 302.22$

Orthorhombic

$P2_12_12_1$

$a = 8.6930$  (5) Å

$b = 12.3768$  (9) Å

$c = 12.4624$  (9) Å

$V = 1340.8$  (3) Å<sup>3</sup>

$Z = 4$

$D_x = 1.497$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 25 reflections

$\theta = 10-13^\circ$

$\mu = 3.0$  mm<sup>-1</sup>

$T = 298$  K

Prism

$0.48 \times 0.20 \times 0.18$  mm

Colorless

Crystal source: crystallized from isopropyl alcohol:tetrahydrofuran (1:1)

### Data collection

Enraf-Nonius CAD-4 diffractometer

$\omega/2\theta$  scans

Absorption correction: empirical

$T_{min} = 0.95$ ,  $T_{max} = 1.00$

4871 measured reflections

3081 independent reflections

2144 observed reflections

$[I > 3\sigma(I)]$

### Refinement

Refinement on  $F$

$R = 0.037$

$wR = 0.037$

$S = 1.549$

2144 reflections

235 parameters

All H-atom parameters refined

$w = 4F_o^2/[\sigma^2(I) + (0.02F_o^2)^2]$

$R_{int} = 0.037$

$\theta_{max} = 27.5^\circ$

$h = -11 \rightarrow 11$

$k = 0 \rightarrow 16$

$l = -16 \rightarrow 16$

3 standard reflections

frequency: 167 min

intensity variation: <2%

$(\Delta/\sigma)_{max} = 0.02$

$\Delta\rho_{max} = 0.87$  e Å<sup>-3</sup>

$\Delta\rho_{min} = -0.42$  e Å<sup>-3</sup>

Atomic scattering factors

from *International Tables for X-ray Crystallography* (1974, Vol. IV) Tables 2.2B, 2.3.1

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$B_{eq} = (8\pi^2/3) \sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$$

	x	y	z	$B_{eq}$
Br1	0.96390 (5)	0.44790 (4)	0.05344 (4)	3.559 (7)
O1	0.7721 (3)	-0.0072 (2)	0.2090 (3)	3.20 (7)
O2	0.9126 (3)	0.1318 (3)	0.2827 (3)	3.25 (7)
N	0.9207 (3)	0.1134 (3)	0.0403 (3)	2.81 (7)
C1	0.7968 (4)	0.1606 (4)	0.1130 (4)	2.51 (9)
C2	0.7831 (4)	0.1061 (3)	0.2201 (4)	2.43 (9)
C3	0.8978 (5)	-0.0526 (4)	0.1493 (4)	3.7 (1)
C4	0.9021 (5)	-0.0074 (4)	0.0394 (4)	3.8 (1)
C5	1.0793 (5)	0.1454 (4)	0.0772 (4)	3.6 (1)
C6	0.8981 (6)	0.1561 (4)	-0.0706 (4)	4.0 (1)
C7	0.6307 (5)	0.1419 (4)	0.2727 (4)	2.84 (9)
C8	0.5004 (5)	0.0752 (4)	0.2572 (4)	3.4 (1)
C9	0.3621 (6)	0.1013 (5)	0.3043 (4)	4.4 (1)

C10	0.3478 (6)	0.1926 (4)	0.3635 (5)	4.5 (1)
C11	0.4698 (6)	0.2601 (4)	0.3749 (4)	4.5 (1)
C12	0.6168 (5)	0.2375 (4)	0.3299 (4)	3.5 (1)
C13	0.7412 (6)	0.3190 (4)	0.3434 (5)	4.9 (1)

Table 2. Selected geometric parameters (Å, °)

O1—C2	1.412 (5)	C3—C4	1.480 (7)
O1—C3	1.437 (5)	C7—C8	1.415 (6)
O2—C2	1.406 (5)	C7—C12	1.387 (7)
N—C1	1.524 (5)	C8—C9	1.377 (6)
N—C4	1.504 (6)	C9—C10	1.355 (8)
N—C5	1.507 (5)	C10—C11	1.357 (7)
N—C6	1.492 (6)	C11—C12	1.423 (7)
C1—C2	1.501 (6)	C12—C13	1.488 (7)
C2—C7	1.543 (6)		
C2—O1—C3	112.8 (3)	C1—C2—C7	108.5 (3)
C1—N—C4	108.1 (3)	O1—C3—C4	110.5 (4)
C1—N—C5	111.4 (3)	N—C4—C3	111.8 (4)
C1—N—C6	108.8 (3)	C2—C7—C8	117.5 (4)
C4—N—C5	111.2 (3)	C2—C7—C12	122.5 (4)
C4—N—C6	109.3 (4)	C8—C7—C12	119.9 (4)
C5—N—C6	108.1 (3)	C7—C8—C9	120.2 (4)
N—C1—C2	114.4 (3)	C8—C9—C10	120.6 (5)
O1—C2—O2	109.5 (3)	C9—C10—C11	119.9 (5)
O1—C2—C1	111.4 (4)	C10—C11—C12	122.7 (5)
O1—C2—C7	105.5 (3)	C7—C12—C11	116.6 (4)
O2—C2—C1	109.2 (3)	C7—C12—C13	125.0 (4)
O2—C2—C7	112.8 (3)	C11—C12—C13	118.4 (4)
C2—O1—C3—C4	-60.7 (5)	N—C1—C2—O2	70.7 (4)
C4—N—C1—C2	48.3 (4)	O2—C2—C7—C12	37.6 (6)
N—C1—C2—O1	-50.3 (4)	H2OH—O2—C2—O1	-29.6 (31)

Refinement of the inversion-related structure under identical conditions yielded:  $R = 0.060$ ,  $wR = 0.069$ ,  $S = 2.828$ . Thus, the absolute configuration of the sample was determined. Programs used were *MolEN* (Fair, 1990) and *ORTEP* (Johnson, 1965).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71435 (23 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HH1076]

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## Patellamide A, a Cytotoxic Cyclic Peptide from the Ascidian *Lissoclinum patella*

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### Abstract

The structure of crystals of patellamide A {13-methyl-9,23-bis(1-methylethyl)-2,16-bis(1-methylpropyl)-14,-28-dioxa-7,21-dithia-3,10,17,24,29,30,31,32-octaazapentacyclo[24.2.1.1<sup>5,8</sup>.1<sup>12,15</sup>.1<sup>19,22</sup>]dotriaconta-1(29),5,-8(30),15(31),19,22(32)-hexaene-4,11,18,25-tetraone methanol solvate monohydrate, C<sub>35</sub>H<sub>49</sub>N<sub>8</sub>O<sub>6</sub>S<sub>2</sub>·CH<sub>4</sub>O·H<sub>2</sub>O}, a cytotoxic cyclic peptide having a non-C<sub>2</sub>-symmetric methyl group, shows the C<sub>2</sub>-symmetric and saddle-shaped rectangular conformation where the methyl group is disordered into two C<sub>2</sub>-symmetric positions. The water and methanol solvents were located on the crystallographic diad axis and were held by hydrogen bonds and van der Waals contacts with the polar ring N atoms and non-polar D-Val side-chain atoms, respectively.

### Comment

As part of a series of studies on the relationship between the chemical structural symmetry and the molecular conformation in cyclic peptides from marine ascidian, the crystal structure of patellamide A (1), a cytotoxic cyclic peptide from *Lissoclinum patella* (Ireland, Durso, Newman & Hacker, 1982) was determined by X-ray single-crystal analysis. The conformational analysis of this molecule, which has a non-C<sub>2</sub>-symmetric methyl group on one side of two C<sub>2</sub>-symmetric dihydro oxazole rings, appears to be important in considering the 'active conformation' of cytotoxic cyclic peptides from ascidian, as seen from the C<sub>2</sub>-symmetric ascidiacyclamide (2) which takes a rectangular conformation (Ishida, Tanaka, Nabae, Inoue, Kato, Hamada & Shioiri, 1988; Ishida, In, Doi, Inoue, Hamada & Shioiri, 1992). The non-C<sub>2</sub>-symmetric patellamide D, in which a benzyl group is attached to only one side, shows a twisted and folded conformation stabilized by four intramolecular